### 841. Advanced Transport Phenomena

Spring, 3(3-0) MTH 215, BS. in engineering or physical science.

Use of equations of change in solving engineering problems. Boundary layer and penetration theories of interphase transport. Potential flow. Theories of turbulence from statistical standpoint.

## Physical Chemistry of 847. Macromolecules

Winter of odd-numbered years. 3(3-0) CHE 446 or approval of department. Interdepartmental with the Department of Chemistry.

Thermodynamics—phase equilibria of polymer solutions; configuration and conformation of chain molecules; characterization of polymer molecular weight and distribution; theoretical and experimental results for dilute solution viscosity and diffusivity; polyelectrolytes.

#### 881. Seminar

Fall, Winter, Spring, Summer. 1(0-2) May reenroll for a maximum of 3 credits allowed toward M.S. degree and 6 credits toward Ph.D. degree.

Detailed library investigation of one or more specialized aspects of chemical engineering, such as recent theoretical developments in one of the unit operations; presentations of these studies to a seminar group. Participation generally required each term of residence.

## 886. Selected Topics in Chemical Engineering

Fall, Winter, Spring, Summer. 3(3-0) May reenroll for a maximum of 9 credits if a different topic is taken.

A newly developing area of chemical engineering selected by the department for offering each term. Information on the specific topic to be covered should be obtained from the department office before registration.

#### 888. Research Survey

Fall, Winter, Spring, Summer. 1 to 3 credits. May reenroll for a maximum of 3

Literature search, problem analysis, and layout of a complete research program.

# Special Problems

Fall, Winter, Spring, S Variable credit. Approval of department. Summer.

#### 899. Master's Thesis Research

Fall, Winter, Spring, S Variable credit. Approval of department. Summer.

# Advanced Chemical Engineering Thermodynamics II 912.

Spring of even-numbered years, 3(3-0) Approval of department.

Relation of thermodynamics to quantum theory and statistical mechanics. Computation of chemical engineering thermodynamic data from spectral measurements. Irreversible spectral measurements. thermodynamics.

### 918. Advanced Chemical Reaction Engineering II

Fall of odd-numbered years. 3(3-0) Approval of department.

Quantitative treatment of current literature in chemical kinetics and reaction engineering.

### 999. **Doctoral Dissertation** Research

Fall, Winter, Spring, S Variable credit. Approval of department. Summer.

# **CHEMISTRY**

**CEM** 

# College of Natural Science

Credit cannot be earned in more than one course of each of the following groups: 130 and 141, and 151; 131, 141, and 151; 132, 241, and 352; 243 and 354; 244 and 354; 245 and 353; 361 and 383; 384 and 461; 385 and 363; 394 and 472.

With department approval, students with advanced placement credit in CEM 151 and 161 may enroll in CEM 181H and 184H. Those with advanced placement credit in CEM 152 may enroll in CEM 182H, and those with advanced placement credit in CEM 153 may enroll in CEM 183H. CEM 181-182-183 is a more advanced treatment of material in CEM 151-152-153. CEM 184-185-186 is a more advanced treatment of material in CEM 151-152-153. Students with credit in an honors chemistry course may not enroll in the corresponding nonhonors course.

#### 130. Introductory Chemistry I

Fall, Winter, Spring, Summer, 4 credits—Self-instructional only. MTH 108 or MTH 111 or concurrently.

Atomic and molecular structure; stoichiometry; gases, liquids, and solids; changes of state.

#### 131. Introductory Chemistry II

Fall, Winter, Spring, Summer, 3 Self-instructional only. CEM 130. Continuation of CEM 130. Chemical kinetics; solutions; acids and bases; equilibria.

# 132. Introductory Chemistry: Carbon Compounds

Fall, Spring, Summer. 4(3-3) CEM 131 or CEM 141 or CEM 151, 139; CEM 161 or CEM 184H.

Chemistry of carbon compounds, introducing the aliphatic and aromatic hydrocarbon series. Some typical compounds are prepared and their behavior studied.

#### 139. Selected Topics in Introductory Chemistry

Fall, Winter, Spring, Summer. 1 to 3 credits. May reenroll for a maximum of 7 credits. Previous college chemistry, approval of credits. department.

Self-instructional units from CEM 130, CEM 131 (or equivalent) selected and approved by the department for individual students with special needs. Each credit requires completion of 9 self-instructional units and the appropriate examinations.

### Introduction to Chemical 141. Principles

Fall, Winter. 4(4-0) MTH 108 or MTH 111 or concurrently; I year high school chemistry and satisfactory chemistry placement test score. Atomic and molecular structure; stoichiometry; gases, liquids, and solids; changes of state; solutions; chemical kinetics; acids and bases; equilibria.

#### 142. Introductory Chemistry III

Fall, Spring. 3(3-0) CEM 131 or CEM 141 or CEM 151, 139.

Reactions and behavior of inorganic compounds illustrated by applications in environmental chemistry.

# Principles of Chemistry I

Fall. 4(4-0) MTH 108 or MTH 111 or concurrently; I year high school chemistry and satisfactory chemistry placement test score or CEM 139 or concurrently.

First of a 3-term sequence for science majors, chemical engineering students, and others desiring a comprehensive general chemistry sequence. Atomic and molecular structure; stoichiometry; solids, liquids, and gases; solutions.

#### Principles of Chemistry II 152.

Winter, Spring, 3(3-0) MTH 112 or concurrently; CEM 151 recommended or CEM 131 or CEM 141 or CEM 181H.

Continuation of CEM 151. Chemical thermodynamics; kinetics, acids, bases, and aqueous equilibria; electrochemistry.

### 153. Introductory Inorganic Chemistry

Fall, Spring. 3(3-0) CEM 152 or CEM 182H.

Continuation of CEM 152. Descriptive inorganic chemistry with further discussion of bonding.

### 161. Introductory Chemistry Laboratory

Fall, Winter, Spring, Summer. 1(0-3) CEM 131 or CEM 141 or CEM 151 or concurrently.

Laboratory work in chemistry, including quantitative physiochemical or analytical quantitative physiochemical or experiments and chemical synthesis.

#### 162. Quantitative Analysis

Fall, Winter, Spring, Summer. 3(1-6) CEM 131 or CEM 141 or CEM 151 or CEM 181H; CEM 161 or CEM 184H. Laboratory work in quantitative chemistry.

## 163. Introductory Inorganic Laboratory

Spring 2(0-6) CEM 162.

Qualitative analysis and inorganic preparations.

## 181H. Honors Chemistry I-Principles

Fall. 4(4-0) An A average in high school chemistry, physics and mathematics; MTH 112 or MTH 122 concurrently. Results of examination during orientation; approval of denartment.

Subatomic, atomic and molecular structure; quantum theory and bonding; experimental methods of structure determination; states of matter; nuclear chemistry.

## 182H. Honors Chemistry 11-Principles

Winter, 4(4-0) CEM 181H with grade of 3.0 or better and/or approval of department. MTH 113 or MTH 123 concurrently.

Kinetic theory of gases, thermodynamics, chemical equilibrium, electrochemistry, chemical kinetics, properties of solutions, macromolecular chemistry.

# Honors Chemistry III—Inorganic Chemistry 183H.

Spring. 3(3-0) CEM 182H with grade of 3.0 or better and/or approval of department. Descriptive inorganic chemistry by periodic groups of elements. Nomenclature, bonding, sterochemistry, and reactions of compounds of the representative and transition elements.

Courses

## 184H. Honors Chemistry Laboratory I

Fall. 1(0-3) CEM 181H concurrently; approval of department.

Techniques of measurement; errors and significant figures; experiments related to atomic and molecular structure.

## Honors Chemistry 185H. Laboratory II

Winter. 2(0-6) CEM 184H; CEM 182H concurrently; approval of department.
Experiments related to gas behavior, thermodynamics, electro-chemistry, chemical

# 186H. Honors Chemistry Laboratory III

kinetics and properties of solutions.

Spring. 2(0-6) Approval of department. Introductory independent laboratory work in chemistry.

#### 241. Organic Chemistry

Fall, Winter, Summer. 4(4-0) CEM 131 or CEM 141 or CEM 151, 139 or CEM 181H; CEM 161 or CEM 184H.

Common classes of organic compounds with emphasis on nomenclature, structural principles, reactions and reaction mechanisms.

#### Organic Chemistry 242.

Winter, Spring, Summer. 4(4-0) CEM

241 Continuation of CEM 241 with emphasis on polyfunctional compounds, particularly groups of compounds having biological significance.

## Organic Chemistry 243. Laboratory

Fall, Winter, Summer. 1(0-2) CEM 241

Introduction to standard organic laboratory techniques.

## Organic Chemistry 244. Laboratory

Winter, Spring, Summer. 1(0-3) CEM 241, CEM 243, CEM 242 concurrently. Organic preparations and qualitative analysis.

#### 245. Organic Chemistry

Fall, Spring. 4(4-0) CEM 242.

Selected topics of organic chemistry, especially compounds of biological interest, discussed with emphasis on mechanisms and stereochemistry. Topics include polymers, amino acids, proteins, sugars, terpenes, steroids, and alkaloids.

#### Instrumental Methods 333.

Spring. 4(2-6) CEM 132 or CEM 241 or CEM 351; CEM 162.

Principles, applications of separation and instrumental analysis. Atomic emission, absorption, fluoresence spectrometry; UV, visible, IR spectrophotometry; molecular fluorescnee; gas and other chromotography; electro-analytical chemistry; electrophoresis; radiochemistry.

#### *351*. Organic Chemistry

Fall. 3(4-0) CEM 152 or CEM 182H.

comprehensive introduction to the fundamentals of organic chemistry designed for chemistry majors but open to others who desire a rigorous, modern treatment of the subject.

# 352. Organic Chemistry Winter. 3(4-0) CEM 351.

Continuation of CEM 351.

#### 353. Organic Chemistry

Spring. 3(4-0) CEM 352. Continuation of CEM 352.

#### 354. Organic Chemistry Laboratory

Winter. 2(0-6) CEM 162 or CEM 185H; CEM 351.

A laboratory course in modern techniques of organic chemistry, including qualitative organic

### 355. Organic Chemistry Laboratory

Spring. 2(0-6) CEM 352, CEM 354. Continuation of CEM 354.

## 356. Organic Chemistry Laboratory

Fall. 2(0-6) CEM 355. Continuation of CEM 355.

#### 361. Chemical Thermodynamics

Fall. 3(4-0) One year general chemistry; one year general physics; MTH 215.

one year general physics, MITA 213. Thermodynamics. Properties of gases. Laws of thermodynamics, properties of ideal and non-ideal solutions, thermodynamics of chemical reactions, activities in non-ionic systems.

## 362. Analytical-Physical Chemistry I

Minter, 3(4-0) CEM 361.

Applications of thermodynamics. Activity coefficients, ionic solutions, cell potentials, ionic equilibria including acid-base, complexation, solubility and redox equilibria, phase equilibria, distillation, extraction, chromatography.

## 363. Analytical-Physical Chemistry II

Spring. 3(4-0) CEM 362.

Chemical kinetics. Homogeneous kinetics, reaction mechanisms, temperature dependence of reaction rates, transport process, heterogeneous kinetics, electrode kinetics, X-ray diffraction, crystal structure.

# Analytical-Physical Chemistry Laboratory I

Winter. 2(1-3) CEM 162; CEM 383, or

Measurement techniques. Temperature measurement and control, pressure, caloimetry, pH, acid-base titrations, cell potentials, treatment of data. CEM 361.

## Analytical-Physical 373. Chemistry Laboratory II

Spring. 2(1-3) CEM 372.

Instrumental measurements. Electrode potentials, chromatography, spectrophotometry, electrolytic conductance, solution kinetics.

### 383. Physical Chemistry: Introductory

Fall, Summer. 3(4-0) CEM 132 or CEM 241 or CEM 351; MTH 113.
Classical and chemical thermodynamics. Introduction to the laws and their applications in treating chemical reactions, pure substances, ideal and non-ideal mixtures, and colligative properties.

### Physical Chemistry: 384. Introductory

Spring. 3(4-0) CEM 383.

Atomic and molecular structure. Atomic and molecular orbitals and chemical bonding. Rotational, vibrational and electronic spectra, nuclear magnetic resonance and electron spin resonance.

## 385. Physical Chemistry: Introductory

Winter. 3(4-0) CEM 383.

Electrochemistry and electromotive Chemical kinetics. Macromolecules biochemical systems. Nuclear chemistry. force.

#### 394. Spectroscopy Laboratory

Spring. 2(1-3) CEM 384 or CEM 461.

Laboratory work in electronic, vibrational, and rotational spectroscopy, mass spectrometry, nuclear and electron spin resonance, dipole moments and magnetic susceptibility.

# 400H. Honors Work

Fall, Winter, Spring, Summer. Variable credit. Seniors, approval of department. Assigned reading and investigation in chemistry under the supervision of the staff. The program will include some creative work.

#### 411. Inorganic Chemistry I

Fall. 3(3-0) CEM 385 or CEM 363.

Principles of structure and bonding in inorganic chemistry, crystal symmetry, coordination chemistry, solvent systems, hydrogen bonding and selected examples from the chemistry of various elements. various elements.

## 412. Inorganic Chemistry II

Winter. 3(3-0) CEM 411.

Inorganic chemistry viewed in a variety of examples of reactions, structure, mechanisms, etc., from representative main group elements and transition elements.

#### 419. Problems and Reports

Fall, Winter, Spring, Summer. 2 to 8 credits.

430. Introduction to Radioactivity

and Radioisotope Techniques

Spring, Summer. 2(3-0) or 3(3-0) One year each of general college chemistry and physics. Interdepartmental with and administered by the Department of Physics.

First 7 weeks. Elementary nuclear processes and properties with emphasis on radioactivity, its measurement, and its interaction with matter. Effects of radiation on chemical and biological systems. Applications of nuclear technology, safety and environmental factors. Last 3 weeks. Fundamentals of nuclear models, reactions and decay mechanisms. Basic principles of nuclear reactors and accelerators.

# Laboratory for Radioactivity and Radioisotope Techniques 431.

Spring, Summer. 1(0-3) CEM 161, PHY 430, concurrently. CEM 162 recommended. Interdepartmental with and administered by the Department of Physics.

Introduction to nuclear instrumentation. Experimental techniques for application of radioisotopes to problems in chemistry, the life sciences, and industry.

#### 461. Theoretical Chemistry I

Fall. 3(4-0) One year general chemistry; one year general physics; MTH 215.

Quantum chemistry. Wave properties, postulates of quantum mechanics, hyrdodgen atom, helium atom, orbital theories, ionic bonds, simple molecules, valence-bond and molecular-orbital theories, complex molecules, introduction to spectra.

#### 462. Theoretical Chemistry II

Winter, 3(4-0) CEM 361, CEM 461.

Spectroscopy and molecular structure. Electronic. infrared, Raman, and microwave spectroscopy, magnetic susceptibility and magnetic resonance, statistical mechanics, statistical thermodynamics, kinetic theory of gases, absolute rate theory.

# Analytical-Physical Chemistry Laboratory III Fall. 2(0-6) CEM 363, CEM 373.

Kinetics, operational amplifiers, polarography, coulometry, electrochemical kinetics, stoppedflow kinetics, digital measurements, neutron activation

## 472. Analytical-Physical Chemistry Laboratory IV

Winter, 2(0-6) CEM 461, CEM 471.

Molecular properties. Mass spectrometry, nuclear and electron spin resonance spectroscopy, infrared spectroscopy, dipole moments, magnetic susceptibility, gaseous decomposition kinetics.

# Modern Physical Chemistry

Spring. 3(3-0) May reenroll for a maximum of 6 credits if a different topic is taken CEM 462.

Topics may be selected from the following: physical properties and structure, molecular structure, spectroscopy, theory of solutions.

## 499. Seminar on Chemical Physics

Fall, Winter, Spring, 1(1-0) May reenroll for a maximum of 3 credits. One year of analytical-physical chemistry. MTH 215; PHY

Literature of chemical physics through oral reports on selected journal articles in the area.

### Advanced Inorganic 810. Chemistry

(812.) Fall. 3(3-0) Approval of

department. Structure, bonding and reactivity patterns of inorganic compounds, with emphasis on metallic elements and coordination chemistry in aqueous and nonaqueous media; concepts in symmetry,

pseudo-rotation and other reaction mechanisms.

## 811. Symmetry, Group and MO Theory

Winter. 3(3-0) Approval of department. Applications of group and molecular orbital theory to chemical bonding, structure and reactions.

## Advanced Inorganic 813. Chemistry—Metals

Spring. 3(3-0) CEM 811.

Continuation of CEM 811 with emphasis on the structure and chemistry of the metals

# Nuclear and Radiochemistry

Winter, 3(3-0) Approval of department. Chemistry of production, isolation and identification of radionuclides and their uses in chemical research.

#### 833. Analytical Spectroscopy

Winter of odd-numbered years, 3(3-0) Approval of department.

#### Advanced Analytical 834. Chemistry

Fall. 3(3-0) Approval of department. Consideration of principles and equilibria pertaining to aqueous and non-aqueous neutralization, redox and complexation reactions and the various separation techniques employed in analyses.

#### 835. Spectrochemical Methods of Analysis

Winter of even-numbered years, 3(2-4) Approval of department.

Principles and applications of atomic absorption, emission. fluorescence: arc and spark emission spectroscopy; UV, visible, IR spectrophotometry; spectrophotometric titrations; reaction rate methods: molecular fluorescence, phosphorescence spectrometry; other optical spectrometric methods.

#### 836. Separations

Spring of odd-numbered years. 3(3-0) Approval of department.

Physical and chemical methods of separation.

#### Electroanalytical Chemistry 837.

Spring of even-numbered years, 3(3-0) Approval of department.

Theory and applications of modern electroanalytical chemistry to chemical and biomedical problems. Coulometry, electrometric titrations, ion-selective voltammetry; electrochemical synthesis and preparation of species for spectroscopy; trace analysis.

#### 838. Scientific Instrumentation

Fall, Spring Summer. 2(1-3) to 4(2-6) May reenroll for a maximum of 8 credits. Approval of department.

Scientific measurements. Principles and applications of servo systems, operational amplifiers, linear and digital solid state devices, analog, digital and hybrid instrumentation systems, and minicomputers for scientific measurements.

### 844. Structural Elucidation by Instrumental Methods

Spring. 3(3-0) Approval of department. A practical instrumental analysis course with the major emphasis on the interpretation of data rather than a detailed description of the instrumentation. The fundamental principles behind the various measurements will be discussed in a control way and impact to the control of the property of the control of the principles will be discussed in a control way and impact to the control of the contr discussed in a general way, and important instrumental limitations will be noted.

#### 847. Physical Chemistry of Macromolecules

Winter of odd-numbered years. 3(3-0) CEM 446 or approval of department. Interdepartmental with and administered by the Department of Chemical Engineering.

Department of Chemical Engineering.

Thermodynamics—phase equilibria of polymer solutions: configuration and conformation of chain molecules; characterization of polymer molecular weight and distribution; theoretical and experimental results for dilute solution viscosity and diffusivity; polyelectrolytes.

## 860. Organic Reactions: A Mechanistic Approach

Fall. 3(3-0) CEM 353; CEM 462 or

approval of department.

Organic reactions are presented in a mechanistic framework. Reactions which proceed via carboncations, carbanions, free radicals, carbenes, arynes and other reactive intermediates, and concerted reactions are included.

#### 861. Structure of Organic Compounds

(851.) Winter. 3(3-0) CEM 860 or

approval of department.

Structural and stereochemical principles will be developed and illustrated. Spectroscopic data will be used to illustrate the principles and to determine structure, with an emphasis on nuclear magnetic resonance spectroscopy.

## 862. Advanced Synthetic Organic Chemistry

(853.) Spring. 3(3-0) CEM 860 or approval of department.

The strategy and methods of organic synthesis will be discussed.

#### 880. Atomic and Molecular Structure

Fall. 3(3-0) CEM 462 or approval of department.

Basic concepts of non-relativistic quantum mechanics will be developed and employed in a descritption of atomic and molecular structure.

#### 881. Thermodynamics

Winter, 3(3-0) Approval of department. Laws of thermodynamics and their application to pure substances and solutions.

#### Chemical Kinetics 883.

Spring. 3(3-0) CEM 880.

Rates and mechanisms of chemical reactions, reaction rate theory, kinetic theory of gases, photochemistry.

## 890. Graduate Problems and Reports

Fall, Winter, Spring, Summer. Variable credit, May reenroll for a maximum of 12 credits. Approval of department.

#### 899. Master's Thesis Research

Fall, Winter, Spring, S Variable credit. Approval of department. Summer. Research in inorganic, analytical, organic, and physical chemistry.

## Selected Topics in Inorganic 913. Chemistry

Fall, Spring, 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. maximum of yerraits if aufferent topic is taken. Rare earth elements, recent advances in the chemistry of metals or nonmetals, high-temperature chemistry. Coordination chemistry and nonaqueous solvents.

## 918. Seminar in Inorganic Chemistry

Fall, Winter, Spring. reenroll for a maximum of 3 credits. 1(1-0) May

Discussions of recent advances and reports by graduate students on research problems.

## 924. Selected Topics in Analytical Chemistry

Fall, Winter, Spring, 3(3-0) or 2(2-0) May reenroll for a maximum of 9 credits if different topic is taken.

Among topics which may be discussed are: advances in electroanalytical chemistry or spectroscopy; nonaqueous solvents; complexation equilibria; surface chemistry; analytical chemistry of polymers.

## 938. Seminar in Analytical Chemistry

Fall, Winter, Spring. reenroll for a maximum of 3 credits. 1(1-0) May Discussions of recent advances and reports by graduate students on research problems.

## Selected Topics in Organic 956. Chemistry

Fall, Winter, Spring. 2(2-0) or 3(3-0) May reenroll for a maximum of 12 credits if different topic is taken. Approval of department. Topics may be selected from heterocyclic chemistry, natural products, free radicals, carbonium ions, organic sulfur or nitrogen compounds, acidity functions, isotope effects, photochemistry and others.

# Courses

## Seminar in Organic 958. Chemistry

Fall, Winter, Spring. 1(2-0) May reenroll for a maximum of 3 credits. Discussions of recent advances and reports by graduate students on research problems.

#### 985. Statistical Thermodynamics

Winter of even-numbered years. Spring. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Partition functions, spectroscopic measurements and thermodynamic applications. Nonequilibrium statistical mechanics and thermodynamics. Time correlation functions and spectroscopic lineshapes, light scattering, and magnetic relaxation. Transport properties of fluids and assets. fluids and gases.

### 987. Selected Topics in Physical Chemistry

Fall, Winter. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Mathematical preparation for quantum chemistry. Selected topics as: kinetics and photochemistry, macromolecular and surface chemistry, molecular spectroscopy, electro and magnetic properties of matter, application of statistical mechanics to chemical problems.

### 988. Selected Topics in Physical Chemistry

Winter, Spring 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department

Topics may be chosen from analysis and interpretation of the spectra of molecules, advanced molecular structure, magnetic resonance, spectroscopy, X-rays and crystal structure, statistical mechanics.

#### 991. Selected Topics in Quantum Chemistry

Fall, Winter. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Approvated apparatum mechanics and application to chemical problems. Selected topics from spectroscopy, properties of atoms and molecules in electric and magnetic fields, and theories of molecular electronic structure.

### 998. Seminar in Physical Chemistry

Fall, Winter, Spring 1(1-0) May reenroll for a maximum of 3 credit.
Discussions of recent advances and reports by graduate students on research problems.

## 999. **Doctoral Dissertation** Research

Fall, Winter, Spring, Summer. Varible credit. Approval of department. Research in analytical, inorganic, organic, and physical chemistry.

# CHINESE

See Linguistics and Oriental and African Languages.

# CIVIL AND SANITARY ENGINEERING

# College of Engineering

# Civil Engineering

CE

#### *251*. Elementary Surveying

Fall, Spring. 4(3-3) Not open to majors. Use of the tape, compass, level, and transit with simple maps; traverse closure and area computations. Profile, cross section and stadia surveys, U.S. land system.

# 252.

Fall, Spring. 5(4-3) Engineering majors or approval of department.

Instruments, theory of measurements, error analysis, stadia, horizontal and vertical curves, U.S. Public Land System, observation for meridian.

### 280. Introduction to Environmental Engineering

Fall, Winter, Spring. 4(4-0) CEM 141, or CEM 131, MTH 112, CPS 120.

Hydrology; ground water and surface water supply systems; wastewater treatment, methods of pollution control for solid waste, air, and noise.

#### 305. Structural Mechanics I

Winter, Spring. 4(4-0) MMM 211.

Stability and determinacy of structures. Two and three dimensional determinate structures. Indeterminate structural analysis by displacement and force methods based upon equilibrium, compatibility and load-deformation

#### 308. Engineering Materials I

Winter, Spring. 4(3-3) MMM 211 or concurrently.

Structure; composition; physical, mechanical and rheological properties of non-metallic construction materials. Emphasis on aggregates, asphalt, inorganic cements, concrete, and wood.

#### 311. Urban Utilities

Winter of odd-numbered years. 3(3-0) Capacities, limitations and cost of public and semi-public utilities as they relate to the planning and design of the urban environment. Topics include transportation, water supply, storm drainage, sewage collection and treatment, solid waste and municipal finance.

#### 312. Soil Mechanics I

Spring, Summer. 4(3-3) MMM 211. Engineering properties of soils and their measurement. Effective stress concept; permeability; fluid flow in soils; stress-strain behavior; soil strength; compaction and consolidation of soils; field exploration and design problems.

### 321. Introductory Fluid Mechanics

Fall, Winter, Spring. 5(4-2) MMM 306. Fluid properties; hydrostatics; control volume approach to conservation of mass, momentum and energy; dimensional analysis and dynamic similitude; fluid resistance; pipe and open channel flows; boundary layer concepts.

### 342 Survey of Transportation Systems

Fall. 4(4-0) Juniors; not open to majors. Survey of engineering aspects of all forms of transportation with emphasis on highway transportation including highway systems, planning, economic and financial aspects, geometrics and traffic studies.

#### 346. Transportation

for system selection.

Winter, Spring, Summer. 3(3-0) MTH

Planning, design and evaluation of transportation systems. Operational characteristics of transportation modes, traffic flow and techniques

#### 347. **Transportation Facilities**

Fall, Winter. 4(3-3) C E 252

Geometric design of highways and airports as these considerations affect capacity, contruction costs, financing and safety.

#### 353. Surveying II

Spring. 4(3-3) C E 251 or C E 252.
Continuation of C E 252 including photogrammetric methods, astronomical observations for latitude, longitude and meridian. Introduction to geodetic methods.

## 370. Cost and Optimization Engineering

Fall, Winter. 3(3-0) MTH 113.

Formulation of engineering decisions governed by current and future costs and returns. Comparison and optimization of alternative engineering projects, products and processes.

#### 372. Construction Estimating

Fall, Spring., 3(3-0) Juniors.

Cost studies of contruction activities with emphasis on labor productivity and operating characteristics of equipment under various site conditions. Interpretation of drawings and specifications.

# Legal Aspects of Engineering

Spring. 3(3-0) Juniors.

The professional engineer's relationship with the legal aspects of engineering. Special emphasis on contract documents.

#### 390. Civil Engineering Analysis

Fall, Spring. 4(4-0) MTH 215 and CPS

Analysis of civil engineering problems by numerical and statistical methods. Approximate methods and error analysis. Application to

#### 400. Structural Mechanics II

computer use.

Fall, Summer. 4(4-0) C E 305.

Miscellaneous topics in displacement calculation by virtual work. Matrix formulation of the general principles of framed structural analysis. Exhaustive study of the flexibility and stiffness methods.

# 405. Structural Design in Steel Fall, Winter. 4(4-0) C E 305.

Beams, columns, tension and compression members, connections. Elastic, plastic and ultimate strength concepts.

### 406. Structural Design in Concrete

Winter, Spring. 4(4-0) CE 305, CE

Reinforced concrete beams, columns, slabs, footings and retaining walls. Elastic theory and ultimate strength concepts. Prestressed theory and design.