# **CHEMISTRY**

# CEM

# **Department of Chemistry** College of Natural Science

### **General Chemistry**

Fall, Spring. 4(4-0) P:M: ((MTH 103 or concurrently) or (MTH 110 or concurrently) or (MTH 116 or concurrently) or (MTH 124 or concurrently) or (MTH 132 or concurrently) or (MTH 152H or concurrently) or (LBS 117 or concurrently) or (LBS 118 or concurrently)) or designated score on Mathematics Placement test Not open to students with credit in CEM 151 or CEM 181H or LBS 171

Elements and compounds; reactions; stoichiometry; thermochemistry; atomic structure; chemical bonding; states of matter; solutions; acids and bases; aqueous equilibria.

#### 142 **General and Inorganic Chemistry**

Fall, Spring. 3(4-0) P:M: CEM 141 or CEM 151 or CEM 181H or LBS 171 Not open to students with credit in CEM 152 or CEM 182H or LBS 172.

Kinetics; gaseous equilibria; acids and bases; pH; buffers; hydrolysis; titrations; heterogeneous equilibria; thermodynamics; redox and electrochemistry; transition metal chemistry; nuclear chemistry; main group chemistry.

#### **Survey of Organic Chemistry** 143

Fall, Spring. 4(3-3) P:M: CEM 141 or CEM 151 Not open to students with credit in CEM

Chemistry of carbon compounds. Chemistry of the main organic functional groups with applications to everyday life, industry, and biology.

#### 151 **General and Descriptive Chemistry**

Fall. 4(4-0) P:M: ((MTH 116 or concurrently) or (MTH 124 or concurrently) or (MTH 132 or concurrently) or (MTH 152H or concurrently) or (LBS 117 or concurrently) or (LBS 118 or concurrently)) or designated score on Mathematics Placement test Not open to students with credit in CEM 141 or CEM 181H or LBS 171.

Atomic structure, chemical bonding and molecular structure; solid state; main group chemistry; acids and bases; transition metal chemistry; coordination chemistry and theories of bonding.

#### 152 **Principles of Chemistry**

Spring. 3(4-0) P:M: CEM 151 or CEM 181H or LBS 171 Not open to students with credit in CEM 142 or CEM 182H or LBS 172.

The mole concept and stoichiometry; solution stoichiometry; thermochemistry; gases, liquids, and solids; kinetics; chemical equilibria; acid-based equilibria; aqueous equilibria; thermodynamics; redox and electrochemistry.

#### 161 Chemistry Laboratory I

Fall, Spring. 1(0-3) P:M: (CEM 141 or concurrently) or (CEM 151 or concurrently) Not open to students with credit in LBS 171L or CFM 185H

Experiments in general chemistry; stoichiometry, calorimetry, electrochemistry, molecular geometry, gas laws, kinetics, acids and bases, and inorganic chemistry.

#### 162 **Chemistry Laboratory II**

Fall, Spring. 1(0-3) RB: (CEM 161 or LBS 171L or CEM 185H) and ((CEM 142 or concurrently) and (CEM 152 or concurrently)) Not open to students with credit in LBS 172L or CEM 186H.

Analytical and inorganic chemistry; redox and acid base titrations; spectrophotometric and gravimetric analysis; preparation and analysis of coordination complexes of nickel, iron, and cobalt.

#### Honors Chemistry I 181H

Fall. 4(4-0) P:M: (MTH 124 or concurrently) or (MTH 132 or concurrently) or (MTH 152H or concurrently) or (LBS 118 or concurrently) R: Approval of department.

Elements and compounds; stoichiometry; reactions; atomic structure and quantum mechanics, chemical bonding and molecular structure; spectroscopy; coordination chemistry and theories of bonding; structure of biochemical molecules.

## **Honors Chemistry II**

Spring. 4(4-0) P:M: (CEM 151 or CEM 181H or LBS 171) and ((MTH 126 or concurrently) or (MTH 133 or concurrently) or (MTH 153H or concurrently)) R: Approval of department.

Thermodynamics and chemical equilibria; acids and bases; redox chemistry; main group elements; solid state; group theory and symmetry; molecular orbital theory; transition metal chemistry and spectroscopy.

### Honors Chemistry Laboratory I

Fall. 2(0-6) P:M: CEM 181H or concurrently R: Approval of department.

Spectroscopy and diffraction methods for the study of electronic structure and molecular geometry; synthesis and separation methods for the preparation and characterization of molecules; application to inorganic, organic, and biochemical molecules and

# Honors Chemistry Laboratory II

Spring. 2(0-6) P:M: CEM 182H or concurrently R: Approval of department.

Laboratory research.

Organic Chemistry I Fall, Spring. 3(4-0) P:M: CEM 141 or CEM 151 or CEM 181H or LBS 171 Not open to students with credit in CEM 351.

Common classes of organic compounds including their nomenclature, structure, bonding, reactivity, and spectroscopic characterization.

# Organic Chemistry II

Fall, Spring. 3(4-0) P:M: CEM 251 Not open to students with credit in CEM 352.

Continuation of CEM 251 with emphasis on polyfunctional compounds, particularly those of biologi-

#### 255 **Organic Chemistry Laboratory**

Fall, Spring. 2(1-3) P:M: (CEM 252 or concurrently) and (CEM 161 or LBS 171L or CEM 185H) Not open to students with credit

Preparation and qualitative analysis of organic compounds.

#### **Quantitative Analysis** 262

Fall, Spring, Summer. 3(3-3) P:M: CEM 162 or LBS 172L Not open to students with credit in CEM 186H.

Preparation and quantitative analysis of chemical compounds.

#### 332 **Instrumental Methods**

Spring. 2(2-0) P:M: (CEM 143 or CEM 251 or CEM 351) and (CEM 162 and MT 213 and MT 417) Not open to students with credit in CEM 333 or CEM 434.

Principles of instrumental analysis and separation techniques.

#### **Instrumental Methods and Applications** 333

Spring. 3(2-3) P:M: {((CEM 143 or CEM 251 or CEM 351) or completion of Tier I writing requirement) and (CEM 262 or CEM 186H)} or (CEM 162 and MT 213 and MT 417) Not open to students with credit in CEM 332.

Principles and applications of instrumental analysis of separation techniques.

#### 351 Organic Chemistry I

Fall. 3(4-0) P:M: CEM 152 or CEM 182H or CEM 142 or LBS 172 Not open to students with credit in CEM 251.

Structure, bonding, and reactivity of organic mole-

#### 352 Organic Chemistry II

Spring. 3(4-0) P:M: CEM 351 Not open to students with credit in CEM 252.

Carboxylate derivatives. Conjugation. Aromaticity. Amino acids. Proteins. Carbohydrates. Nucleic acids

#### Organic Laboratory I 355

Spring. 2(0-6) P.M. (CEM 162 or CEM 186H or LBS 172L or (CEM 352 or concurrently)) or completion of Tier I writing requirement Not open to students with credit in CEM 255.

Organic laboratory techniques. Distillation. Spectroscopy. Melting points. Recrystallization. Chromatography. Measuring physical properties.

#### Organic Laboratory II 356

Fall. 2(0-6) P:M: CEM 355

Multi-step organic synthesis. Qualitative organic analysis. Separation, identification, and characterization of unknowns.

#### Introductory Physical Chemistry I 383

Fall. 3(4-0) P:M: (CEM 142 or CEM 152 or CEM 182H or LBS 172) and (MTH 133 or MTH 153H or MTH 126 or LBS 119) Not open to students with credit in CEM 391.

Physical chemistry of macroscopic systems: thermodynamics, kinetics, electrochemistry.

#### 384 Introductory Physical Chemistry II

Spring. 3(4-0) P:M: CEM 383 Not open to students with credit in CEM 461.

Physical chemistry of microscopic systems: quantum mechanics, spectroscopy.

#### 391 **Molecular Thermodynamics**

Fall. 3(4-0) P:M: (CEM 142 or CEM 152 or CEM 182H) and (MTH 234 or MTH 254H or LBS 220) and (PHY 184 or PHY 232) RB: One year of general chemistry, calculus, and general physics. SA: CEM 361 Not open to students with credit in CEM 383.

Statistical mechanics and its use in classical chemical thermodynamics. Applications of thermodynamics to chemical systems at equilibrium. Introduction to chemical kinetics.

### 392

**Quantum Chemistry**Fall, Spring. 3(4-0) P:M: CEM 391 and (MTH 234 or LBS 220 or MTH 254H) RB: One year of general chemistry, calculus through differential equations, and general physics. SA: CEM 362, CEM 461 Not open to students with credit in CEM 384.

Postulates of quantum mechanics and their application to model systems, atoms and molecules. Introduction to molecular spectroscopy.

#### 395 Analytical/Physical Laboratory

Spring. 2(1-4) P:M: (CEM 391 or CEM 383) and (CEM 262 and completion of Tier I writing requirement) RB: One year of general chemistry, calculus, and general physics. SA: CEM 372, CEM 472

Chemical kinetics, thermodynamics, and computerbased data analysis methods.

#### 400H **Honors Work**

Fall, Spring, Summer. 1 to 12 credits. A student may earn a maximum of 12 credits in all enrollments for this course. P:M: Completion of Tier I writing requirement. R: Approval of department.

Readings and investigations in chemistry.

### Literature and Writing in Chemistry

Spring. 3 credits. P:M: (CEM 252 and CEM 384) and ((CEM 333 or concurrently) and completion of Tier I writing requirement) R: Open only to juniors or seniors in the Chemistry major.

Library research related to a topic in contemporary chemistry; thesis required.

Inorganic Chemistry Spring. 4(4-0) P:M: CEM 383 or CEM 391 Principles of structure and bonding. Symmetry. Solid state chemistry. Acid-base and redox reactions. Main group chemistry: transition metal bonding, spectra, and reaction mechanisms.

## **Advanced Synthesis Laboratory**

Fall. 3(0-8) P:M: (CEM 411) and completion of Tier I writing requirement. R: Open only to juniors or seniors in the Bachelor of Science degree program in Chemistry.

Methods of synthesizing inorganic and organometallic compounds.

#### 419 Independent Study

Fall, Spring, Summer. 1 to 12 credits. A student may earn a maximum of 12 credits in all enrollments for this course. R: Approval of department.

Faculty supervised readings in chemistry.

#### Independent Research 420

Fall, Spring, Summer. 1 to 12 credits. A student may earn a maximum of 12 credits in all enrollments for this course. R: Approval of department.

Faculty supervised independent investigations in

#### **Advanced Analytical Chemistry** 434

Fall. 3(3-0) P:M: CEM 392 and CEM 395 and CEM 352 SA: CEM 361, CEM 362

Instrumental methods of analysis, including spectroscopy, chromatography and electrochemistry.

#### **Analytical Chemistry Laboratory** 435

Spring. 2(1-3) P:M: (CEM 434 or concurrently) and completion of Tier I writing requirement. SA: CEM 372, CEM 472

Electronic and optical components of chemical instrumentation. Spectroscopic and chromatographic methods.

#### 444 **Chemical Safety**

Fall. 1(1-0) P:M: CEM 142 and CEM 252 Prudent laboratory practices. Regulatory agencies' expectations of chemical industries and academia.

### Seminar in Computational Chemistry

Fall, Spring, Summer. 1 to 6 credits. A student may earn a maximum of 6 credits in all enrollments for this course. P:M: (MTH 133 and CSE 231) and ((CEM 152 or concurrently) or (CEM 182H or concurrently)) RB: CPS 260 and CEM 351

Written and oral reports on selected journal articles in computational chemistry.

#### 485 **Modern Nuclear Chemistry**

Spring of even years. 3(3-0) P:M: (CEM 141 or CEM 152 or CEM 182H) and (PHY 232 or PHY 184) RB: CEM 392 or CEM 384 or PHY 471 SA: CEM 430

Elementary nuclear processes and properties; radioactivity, its measurement and its interaction with

#### 495 Molecular Spectroscopy

Fall. 2(1-4) P:M: CEM 392 and CEM 395 RB: One year of physical chemistry. SA: **CEM 472** 

Experiments in magnetic resonance, optical, and vibrational spectroscopies.

#### **Chemical Physics Seminar** 499

Fall, Spring, Summer. 1(1-0) A student may earn a maximum of 2 credits in all enrollments for this course. P.M: ((PHY 321) and completion of Tier I writing requirement) and (MTH 235 or LBS 220 or MTH 255H)

Written and oral reports on selected journal articles in chemical physics.

## Advanced Inorganic Chemistry I

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Principles of chemical bonding, electronic structure, and reaction mechanisms of main group and transition metal compounds. Concepts of group theory.

# Advanced Inorganic Chemistry II

Spring. 3(3-0) RB: CEM 811 R: Open only to graduate students in College of Natural

Science or College of Engineering.

Descriptive chemistry of inorganic compounds.

Emphasis on synthesis, structure, and reactivity patterns of coordination, organometallic, and solid state compounds of transition metals and main group elements.

### 820 Organometallic Chemistry

Organometallic functional groups. Principles of electronic structure, and bonding in organometallic species will be related to reactivity patterns in common systems. Preparation of complexes with applications to catalytic and stoichiometric organic syntheses

#### 832 Mass Spectrometry

Fall, Spring. 3(3-0) R: Open only to graduate students in the College of Natural Science or College of Engineering.

Instrumentation of mass spectrometry. Interpreting mass spectra of organic and inorganic molecules. Applications to analysis of large molecules and chromatography.

#### 834 **Advanced Analytical Chemistry**

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Principles of equilibria and applications in analytical methodology. Acid-base, complexation, redox reactions. Potentiometry and conductometry. Solute partitioning in extraction and chromatography. Kinetic methods of analysis.

#### 835 **Spectrochemical Methods of Analysis**

Spectrochemical Methods of Analysis

Spring of even years. 3(2-3) R: Open only to graduate students in College of Natural Science or College of Engineering.

Principles and applications of atomic absorption, emission, fluorescence. Plasma emission spectros-

copy. UV, visible, IR spectrophotometry. Reactionrate methods. Molecular fluorescence and phosphorescence. Principles and applications of lasers.

#### 836 Separation Science

Spring of odd years. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Physical and chemical principles of separations, column technology, and instrumentation for gas, liquid, and supercritical fluid chromatography.

# 837

**Electroanalytical Chemistry**Fall of even years. 3(3-0) R: Open only to graduate students in College of Natural

Science or College of Engineering.

Modern electroanalytical chemistry. Theory and applications to chemical and biological problems. Coulometry, voltammetry, ion-selective potentiometry, and other electrochemical techniques.

#### 838 Computer-Based Scientific Instrumentation

Fall. 3(1-6) A student may earn a maximum of 6 credits in all enrollments for this course. R: Open only to graduate students in College of Natural Science or College of Agriculture and Natural Resources.

Electronic and computer-aided measurement and control in scientific instrumentation and experimentation. Principles and applications of digital computers, operational amplifiers, digital logic devices, analog-to-digital converters, and other electronic instruments.

### Structure and Spectroscopy of Organic 845 Compounds

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Structural and stereochemical principles in organic chemistry. Applications of spectroscopic methods, especially nuclear magnetic resonance, static and dynamic aspects of stereochemistry. Spectroscopy in structure determination.

#### 850 Intermediate Organic Chemistry

Traditional and modern basic reaction mechanisms and principles and their synthetic applications.

# 851

Advanced Organic Chemistry
Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Structure, reactivity, and methods. Acid-base reactions, substitution, addition, elimination, and pericyclic processes. Major organic intermediates related to simple bonding theory, kinetics, and thermodynam-

#### 852 **Methods of Organic Synthesis**

Spring. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Principal reactions leading to carbon-carbon bond formation and functional group transformations. Strategies and methods of organic synthesis.

#### 881 **Atomic and Molecular Structure**

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Postulates of quantum mechanics, analytical solu-

tions of the Schroedinger equation, theoretical descriptions of chemical bonding, spectroscopy, statistical mechanics, and statistical thermodynamics.

#### 882 **Kinetics and Spectroscopic Methods**

Spring. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Rate equations and mechanisms of chemical reactions: reaction rate theory, kinetic theory of gases, photochemistry. Spectroscopic methods, and applications of spectroscopy in reaction kinetics.

#### 883 **Computational Quantum Chemistry**

Fall. 3(2-3) RB: CEM 461 or CEM 881

Computational methods in determining electronic energy levels, equilibrium nuclear configurations, and other molecular properties.

#### 888 **Computational Chemistry**

Spring. 3(2-3)

Computational approaches to molecular problems. Use of ab initio and semi-empirical electronic structure, molecular mechanics and molecular dynamics software.

#### 890 **Chemical Problems and Reports**

Fall, Spring, Summer. 1 to 6 credits. A student may earn a maximum of 12 credits in all enrollments for this course.

Investigation and report of a nonthesis problem in chemistry.

#### 899 Master's Thesis Research

Fall, Spring, Summer. 1 to 20 credits. A student may earn a maximum of 99 credits in all enrollments for this course. R: Open only to graduate students in the Department of Chemistry.

Master's thesis research.

#### 913 **Selected Topics in Inorganic Chemistry**

Fall, Spring. 1 to 3 credits. A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to graduate students in College of Natural Science or College of Engineering.

Chemistry of metal-metal bonds and clusters, organometallic chemistry, layered oxides, and complex layered oxides. Photochemistry. Solid state chemistry and applications of quantum mechanics.

## 918

Inorganic Chemistry Seminar
Fall, Spring. 1(1-0) A student may earn a maximum of 3 credits in all enrollments for this course. R: Open only to graduate students in the Department of Chemistry.

Advances in inorganic chemistry reported by graduate students.

#### 924 **Selected Topics in Analytical Chemistry**

Fall, Spring. 2 to 3 credits. A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to graduate students in College of Natural Science or College of Engineering.

Advanced computer techniques, surface chemistry, analytical chemistry of polymers, or statistics for chemists.

#### 938 **Analytical Chemistry Seminar**

Fall, Spring. 1(1-0) A student may earn a maximum of 3 credits in all enrollments for this course. R: Open only to graduate students in College of Natural Science or College of Engineering.

Advances in analytical chemistry reported by graduate students, faculty, and guest lecturers.

### **Selected Topics in Organic Chemistry**

Fall, Spring. 1 to 3 credits. A student may earn a maximum of 12 credits in all enrollments for this course. R: Open only to graduate students in College of Natural Science or College of Engineering.

Heterocyclic and organometallic chemistry, natural products, photochemistry, free radicals, or reaction mechanisms.

#### 958 **Organic Chemistry Seminar**

Fall, Spring. 1(1-0) A student may earn a maximum of 2 credits in all enrollments for this course. R: Open only to graduate students in College of Natural Science or Col-

lege of Engineering.
Advances in organic chemistry reported by graduate students.

#### **Emerging Topics in Chemistry** 971

Fall, Spring. 1 to 3 credits. A student may earn a maximum of 6 credits in all enrollments for this course. R: Open only to doctoral students in the Chemistry or Chemical Physics major.

Discussion of a research topic of emerging interest in chemistry. Preparation of a proposal for funding of research.

#### 987 Selected Topics in Physical Chemistry I

Fall. 1 to 3 credits. A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to doctoral students or approval of department.

Topics such as kinetics and photochemistry, macromolecular and surface chemistry, molecular spectroscopy, electric and magnetic properties of matter, or applications of statistical mechanics to chemical problems.

#### Selected Topics in Physical Chemistry II 988

Spring. 1 to 3 credits. A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to doctoral students or approval of department.

Topics such as analysis and interpretation of molecular spectra, advanced molecular structure theory, magnetic resonance, X-rays and crystal structure, scientific analysis of vacuum systems, or problems in statistical mechanics.

#### 991 **Quantum Chemistry and Statistical** Thermodynamics I

Fall. 3(3-0) R: Open only to graduate students in College of Natural Science or College of Engineering.

Principles and applications of quantum chemistry. Partition functions, spectroscopic measurements, and thermodynamic applications.

### **Quantum Chemistry and Statistical** 992 Thermodynamics II

Spring. 3(3-0) RB: CEM 991

Analytical and numerical methods for solving quantum chemical problems. Statistical mechanics of solids and liquids.

#### **Advanced Topics in Quantum Chemistry** 993

Spring of odd years. 3(3-0) A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to graduate students in the College of Natural Science or the College of Engineering.

Spectroscopic theory, properties of atoms and molecules in electric and magnetic fields, intermolecular forces. Many-body theory, molecular electronic structure, solid state chemistry, or molecular reaction dynamics.

#### 994 **Advanced Topics in Statistical** Mechanics

Spring of even years. 3(3-0) A student may earn a maximum of 9 credits in all enrollments for this course. R: Open only to graduate students in the College of Natural Science or the College of Engineering.

Nonequilibrium statistical mechanics and thermodynamics. Correlation functions and spectroscopy, light scattering, magnetic relaxation, transport properties of fluids and gases, or statistical mechanics of chemical reactions.

### 998

Physical Chemistry Seminar
Fall, Spring. 1(1-0) A student may earn a maximum of 3 credits in all enrollments for this course. R: Open only Chemistry. dents in the Department of Chemistry.

Advances in physical chemistry reported by graduate students.

#### 999 **Doctoral Dissertation Research**

Fall, Spring, Summer. 1 to 24 credits. A student may earn a maximum of 120 credits in all enrollments for this course. R: Open only to doctoral students in the Department of Chemistry.

Doctoral dissertation research.