# **CHEMICAL ENGINEERING**

# **Department of Chemical Engineering College of Engineering**

Material and Energy Balances Fall, Spring. 3(4-0) P:M: (MTH 133) and (CEM 142 or CEM 143 or CEM 152) and (CSE 101 or concurrently or CSE 131 or concurrently)

CHE

Chemical engineering calculations. Synthesis of chemical process systems. Analysis of chemical processes using material and energy balances. Enthalpy calculations for changes in temperature, phase transitions, and chemical reactions.

### 301 Chemical Engineering as a Profession Fall. 1(2-0) P:M: (CHE 201 or concurrently)

RB: Junior standing in chemical engineering R: Open only to students in the Chemical Engineering major.

Professional aspects of chemical engineering. Communication skills, professionalism and ethics, teamwork skills, contemporary engineering issues, career planning, project management, industrial

#### 311 Fluid Flow and Heat Transfer

Fall. 4(5-0) P:M: (CHE 201 or concurrently and MTH 235 or concurrently) R: Open only to students in the College of Engineering. Not open to students with credit in ME 201 or MSM 351.

Thermodynamics of fluid flow. Laminar and turbulent flow. Design of flow systems. Heat transfer in solids and flowing fluids. Interphase heat transfer. Radiant heat transfer. Multiple effect evaporation. Design of heat exchange equipment.

### 312 Mass Transfer and Separations

Spring. 4(5-0) P:M: (CHE 201 and MTH 235 or concurrently) R: Open only to students in the College of Engineering.

Diffusion. Mass transfer coefficients. Design of countercurrent separation systems, both stagewise and continuous. Distillation, absorption, extraction. Multicomponent separations. Batch processes. Computer-aided design methods.

## 316

Unit Operations Laboratory Spring. 3(1-6) P:M: (CHE 311 and CHE 312 or concurrently and CHE 321 or concurrently and CHE 431 or concurrently) and completion of Tier I writing requirement. R: Open only to students in the Department of Chemical Engineering.

Momentum, heat, and mass transfer. Separation processes: distillation, filtration, and drying. Reactor kinetics. Automatic process control. Laboratory problems requiring team effort.

### Thermodynamics for Chemical 321 Engineering

Spring. 4(5-0) P:M: (CHE 201)

First and second laws. Thermodynamics of flow and energy conversion processes. Properties of single and multi-component systems. Phase equilibria. Chemical equilibria in reacting systems.

#### Transport Phenomena 422

Spring. 3(3-0) P:M: (CHE 311 and CHE 312)

Mathematical and physical analogies among mass, energy and momentum transfer processes. Dimensional analysis and solutions to multivariable boundary value problems. Numerical solutions to nonlinear problems

Chemical Reaction Engineering Spring. 3(3-0) P:M: (CHE 311 and CHE 312 or concurrently and CHE 321 or concur-

Design and analysis of homogeneous flow and batch reactors. Chemical kinetics and equilibria. Reaction rate expressions from mechanisms and experimental data. Mass and heat transfer in heterogeneous reactors. Heterogeneous reactor design. Catalysis.

432 Process Dynamics and Control Fall. 3(3-0) P:M: (CHE 431) Mathematical modeling of process dynamics. Control theory. Design of control systems and specification of control hardware. Integration of control theory with modern practice.

## Process Design and Optimization I

Fall. 4(5-0) P:M: (CHE 432 or concurrently) and completion of Tier I writing requirement. R: Open only to students in the Department of Chemical Engineering.

Applications of chemical engineering principles in design calculations. Selection of optimum design. Influence of design on capital investment, operating cost, product loss and quality. Mathematical programming methods for optimization.

# Process Design and Optimization II Spring. 2(0-4) P:M: (CHE 433)

Design project requiring an integrated design of chemical engineering processes. Process and project engineering. Instrumentation and control systems. Flowsheet layout and optimization. Process simulation

Composite Materials Processing Fall. 3(2-3) P:M: (CHE 311 or ME 332 or CE 321)

Manufacturing processes for thermoset and thermoplastic matrix composites. Mechanical and thermal evaluation of composites. Rheology and molding of fiber-filled materials.

# Chemical Engineering Principles in Polymers and Materials Systems Spring. 3(3-0) P:M: (CHE 311 and CHE 321 473

and CHE 431 and CEM 352)

Apply chemical engineering principles to polymer and materials systems. Structures and properties of metals, ceramics and polymers. Thermodynamics, synthesis, rubber elasticity, viscoelasticity, kinetics, rheology, and processing of polymers systems. Applications of statistics and problem-solving skills to materials systems.

# **Biochemical Engineering**

Fall. 3(2-3) P:M: (CHE 431)

Applications of microbiology and biochemistry to biochemical engineering. Kinetics and thermodynamics of biochemical reactors. Transport phenomena in biological systems. Bioreactor design and

#### **49**0 Independent Study

Fall, Spring, Summer. 1 to 3 credits. A student may earn a maximum of 6 credits in all enrollments for this course. R: Open only to juniors or seniors or graduate students in the Department of Chemical Engineering. Approval of department.

Theoretical or experimental studies of current research topics in chemical engineering. Individual interaction with faculty adviser.

# **Selected Topics in Chemical Engineering**

Fall, Spring. 1 to 3 credits. A student may earn a maximum of 6 credits in all enrollments for this course. R: Open only to juniors or seniors or graduate students in the Department of Chemical Engineering.

Study of newly developing or non-traditional chemical engineering topics in a classroom environment.

### **Advanced Chemical Engineering** Calculations

Spring. 3(3-0)
Formulation of differential equations modelling physical phenomena in chemical engineering. Application of analytical and numerical solution methods. Interpretation of solutions.

### 804 Foundations in Chemical Engineering I Summer. 3(2-2)

Mass and energy balances in batch, continuous and open systems. Process thermodynamics. Cryogenics. Properties of substances and mixtures. Phase equilibria. Chemical reaction equilibria. Chemical reactor kinetics. Process design orientation.

### Foundations in Chemical Engineering II 805 Summer. 3(2-2)

Momentum, energy, and mass transfer. Laminar and turbulent flow. Fluid friction. Dimensional analysis. Heat transfer in stationary and flowing materials. Interchanges. Condensation. Boiling. Binary and multicomponent distillation, absorption, extraction.

### **Advanced Chemical Engineering** 821 Thermodynamics

Fall. 3(3-0) R: Open only to Chemical Engineering majors.

Laws of thermodynamics, unsteady state processes. Prediction and correlation of phase equilibria for nonelectrolytes. Relation of quantum theory and statistical mechanics to thermodynamic properties.

### **Advanced Transport Phenomena** 822

Fall. 3(3-0) P:NM: (CHE 801)

Derivation of balance equations for mass, energy, and momentum. Constitutive equations for multicomponent fluids. Estimates of transport properties. Approximate models for turbulent and boundary layer flows. Boundary value problems.

### 831 **Advanced Chemical Reaction** Engineering

Fall. 3(3-0)

Characterization of solid catalysts. Heterogeneous reaction rate expressions. Simultaneous mass and heat transport and chemical reaction in porous catalysts. Design of fixed-bed and fluidized-bed reactors. Industrial catalytic reactions.

# Chemical Engineering-CHE

#### 871 Material Surfaces and Interfaces

Fall of odd years. 3(3-0) Interdepartmental with Materials Science and Mechanics. Administered by Department of Materials Science and Mechanics. P:NM: (CEM 362 or MSM 351) R: Open only to graduate students in the Department of Chemical Engineering or Department of Chemistry or Department of Materials Science and Mechanics or School of Packaging.

Physical and chemical nature of solid surfaces and their interaction with gases, liquids, and other solids. Characterization of surfaces and solid-solid interfaces. Relation of surface and interfacial structure to engineering phenomena.

## Polymers and Composites: Manufacturing, Structure and 872 Performance

Spring of even years. 3(3-0) R: Open only to graduate students in the College of Engineering or the Department of Chemistry.

Structure-Property Relations of Polymers, Fibers, Fabrics and Composites, Material Selection, Manufacturing Processes, Process Induced Microstructure, Prediction of Composite Mechanical Properties, Dimensional Stability, Design of Cure Cycles, Mold Design.

### **Advanced Biochemical Engineering** Spring of even years. 3(3-0)

Microbial strain improvement. Metabolic engineering. Structured growth models. Non-ideal bioreactor performance. Biosensors and process control of bioreactors. Separation processes for biochemicals.

#### 890 Independent Study

Fall, Spring, Summer. 1 to 3 credits. A student may earn a maximum of 6 credits in all enrollments for this course. R: Open only to Chemical Engineering majors. Approval of department.

Supervised individual investigation of a problem in chemical engineering.

### 891 Selected Topics

Fall, Spring, Summer. 1 to 3 credits. A student may earn a maximum of 6 credits in all enrollments for this course. R: Open only to Chemical Engineering majors.

Physical and mathematical analysis of phenomena such as swirling flows or stability of reactions and transport processes.

#### 892 Seminar

Fall, Spring. 1(0-2) A student may earn a maximum of 4 credits in all enrollments for this course. R: Open only to Chemical Engineering majors.

Presentations of detailed studies on one or more specialized aspects of chemical engineering.

### Master's Thesis Research 899

Fall, Spring, Summer. 1 to 8 credits. A student may earn a maximum of 24 credits in all enrollments for this course. R: Open only to Chemical Engineering majors.

Master's thesis research.

# Viscoelasticity and Flow of Polymeric

Spring of odd years. 3(3-0)

Time dependent and steady flow properties of polymeric materials related to molecular and structural parameters. Examples of polymeric blends and composites with thermoplastic and thermoset components.

#### **Doctoral Dissertation Research** 999

Fall, Spring, Summer. 1 to 12 credits. A student may earn a maximum of 72 credits in all enrollments for this course. R: Open only to Chemical Engineering majors.

CEM

Doctoral dissertation research.

# **CHEMISTRY**

# **Department of Chemistry** College of Natural Science

**General Chemistry**Fall, Spring. 4(4-0) P:M: (MTH 103 or concurrently or MTH 110 or concurrently or MTH 116 or concurrently or MTH 124 or concurrently or MTH 132 or concurrently or MTH 152H or concurrently or LBS 117 or concurrently or LBS 118 or concurrently) or designated score on Mathematics placement test. Not open to students with credit in CEM 152 or CEM 182H or LBS 171.

Atoms, molecules, ions; chemical calculations; reactions, energy changes; gases; periodic properties of elements; chemical bonds; states of matter, solutions; acids and bases; aqueous reactions and ionic equations.

General and Inorganic Chemistry Fall, Spring. 3(4-0) P:M: (CEM 141 or LBS 171) Not open to students with credit in CEM 151 or LBS 172.

Kinetics; gaseous equilibria; acids and bases; pH; aqueous equilibria involving buffers, hydrolysis, and titrations; heterogeneous equilibria of weakly soluble salts; electrochemistry; coordination chemistry, stereochemistry, and bonding within the transition elements.

## Survey of Organic Chemistry

Fall, Spring. 4(3-3) P:M: (CEM 141 or CEM 151) Not open to students with credit in CEM 251 or CEM 351.

Chemistry of carbon compounds. Chemistry of the main organic functional groups with applications to everyday life, industry, and biology.

## General and Descriptive Chemistry

Fall. 4(4-0) P:M: (MTH 116 or concurrently or MTH 124 or concurrently or MTH 132 or concurrently or MTH 152H or concurrently or LBS 117 or concurrently or LBS 118 or concurrently) or designated score on Mathematics placement test. Not open to students with credit in CEM 142 or CEM 181H or LBS 172

Atomic and molecular structure; ionic and molecular bonding models; periodic trends; chemical reactivity by periodic group; nomenclature, structure, bonding and reactivity of coordination compounds; bioinorganic chemistry.

## **Principles of Chemistry**

Spring. 3(3-0) P:M: (CEM 151) Not open to students with credit in CEM 141 or CEM 182H or LBS 171.

The mole concept; stoichiometry and chemical calculations; gas laws; phase changes; thermodynamics; enthalpy, entropy and free energy; crystal structures; properties of solutions; chemical kinetics; gaseous equilibria; theory and reactions of acids/bases; aqueous equilibria; electrochemistry.

### 161

Chemistry Laboratory I
Fall, Spring. 1(0-3) P:M: (CEM 141 or concurrently or CEM 151 or concurrently) Not open to students with credit in LBS 171L or CFM 185H

Experiments in general chemistry; stoichiometry, calorimetry, electrochemistry, molecular geometry, gas laws, kinetics, acids and bases, and inorganic

#### 162 Chemistry Laboratory II

Fall, Spring. 1(0-3) P:NM: (CEM 161 or LBS 171L or CEM 185H) and (CEM 142 or concurrently and CEM 152 or concurrently) Not open to students with credit in LBS 172L or CEM 186H.

Analytical and inorganic chemistry; redox and acid base titrations; spectrophotometric and gravimetric analysis; preparation and analysis of coordination complexes of nickel, iron, and cobalt.

Honors Chemistry I Fall. 4(4-0) P:M: (MTH 124 or concurrently or MTH 132 or concurrently or MTH 152H or concurrently or LBS 118 or concurrently) R: Approval of department.

States of matter. Descriptive inorganic chemistry by periodic groups of elements. Kinetic theory of gases. Thermodynamics, chemical equilibrium and electrochemistry. Properties of solutions. Macromolecular chemistry. Macroscopic kinetics.

### Honors Chemistry II

Spring. 4(4-0) P:M: (CEM 181H) and (MTH 126 or concurrently or MTH 133 or concurrently or MTH 153H or concurrently) R: Approval of department.

Subatomic, atomic and molecular structure. Quantum theory and bonding. Stereochemistry and ro-menclature. Experimental methods of structure determination. Reactions of compounds of the maingroup and transition elements. Reaction dynamics. Nuclear chemistry.

Honors Chemistry Laboratory I Fall. 2(0-6) P:M: (CEM 181H or concurrently) R: Approval of department.

Spectroscopy and diffraction methods for the study of electronic structure and molecular geometry; synthesis and separation methods for the preparation and characterization of molecules; application to inorganic, organic, and biochemical molecules and materials

Honors Chemistry Laboratory II Spring. 2(0-6) P:M: (CEM 182H or concurrently) R: Approval of department. Laboratory research.

Organic Chemistry I Fall, Spring. 3(4-0) P:M: (CEM 141 or CEM 152 or CEM 182H or LBS 171) Not open to students with credit in CEM 143 or CEM

Common classes of organic compounds including their nomenclature, structure, bonding, reactivity, and spectroscopic characterization.

## Organic Chemistry II

Fall, Spring. 3(4-0) P:M: (CEM 251) Not open to students with credit in CEM 352.

Continuation of CEM 251 with emphasis on polyfunctional compounds, particularly those of biologi-