CHEMISTRY CEM

College of Natural Science

Credit cannot be earned in more than one course of each of the following groups: 141A and 141B and 151; 143, 241, and 351; 142 and 153; 242 and 352; 243 and 354; 244 and 355; 245 and 353; 361 and 383; 363 and 385; 384 and 461.

With department approval, students with advanced placement credit in CEM 151 and 161 may enroll in CEM 181H and 184H. Those with advanced placement credit in CEM 152 may enroll in CEM 182H, and those with advanced placement credit in CEM 153 may enroll in CEM 183H. CEM 181H-182H-183H is a more advanced treatment of material in CEM 151-152-153. CEM 184H-185H-186H is a more advanced treatment of material in CEM 161-162-163. Students with credit in an honors chemistry course may not enroll in the corresponding nonhonors course.

139. Selected Topics in Introductory Chemistry

Fall, Winter, Spring, Summer. 1 to 3 credits. May reenroll for a maximum of 7 credits. Previous college chemistry, approval of department.

Self-instructional units from CEM 140, CEM 141A, CEM 141B (or equivalent) selected and approved by the department for individual students with special needs.

140. Introductory Chemistry

Fall, Winter, Spring, Summer. 2 credits. Self-scheduled instruction only. MTH 108 or MTH 111 or concurrently.

Chemical symbols, formulas, equations, stoichiometry, structure of atoms, bonding, states of matter, solutions.

141A. Chemical Principles

(141.) Fall, Winter, Spring, Summer. Fall 4(4-0); Winter, Spring, Summer: 4 credits. Self-scheduled instruction only. MTH 108 or MTH 111 or concurrently; CEM 140 or satisfactory chemistry placement test score.

Chemical principles for students in the physical sciences and engineering.

141B. Chemical Principles

Fall, Winter, Spring, Summer. Fall 4(4-0); Winter, Spring, Summer: 4 credits. Self-scheduled instruction only. MTH 108 or MTH 111 or concurrently; CEM 140 or satisfactory chemistry placement test score.

Chemical principles for students in biological, health-related, and agricultural disciplines.

142. Descriptive Inorganic Chemistry

Winter, Spring. 3(3-0) CEM 141A or CEM 141B or CEM 152.

Reactions and behavior of inorganic compounds illustrated in part by industrial and environmental applications.

143. Introductory Organic Chemistry

(132.) Fall, Spring, Summer. 4(3-3) CEM 141A or CEM 141B or CEM 152.

Chemistry of carbon compounds, introducing the aliphatic and aromatic hydrocarbon series. Some typical compounds are prepared and their behavior studied.

481. Transport Phenomena

Spring. 3(3-0) CHE 342, CHE 381.

Fundamental treatment of momentum, energy and mass transport. Use of partial differential equations and equations of change for chemical engineering applications. Analogies among the phenomena, dimensional analysis, and boundary layer theory.

801. Advanced Chemical Engineering Calculations I

Fall. 3(3-0) CHE 381.

Chemical engineering applications of advanced mathematical methods. Formulation and solution of mathematical equations which describe physical problems. Computer solutions.

802. Advanced Chemical Engineering Calculations II

Winter. 3(3-0) CHE 801.

Continuation of CHE 801.

806. Thermodynamics and Kinetics in Chemical Engineering

Summer. 5(7-0) B.S. with a major in chemistry, biochemistry, or a closely allied area. Mathematics through calculus. College level physics. General physical, and organic chemistry. Not open to students with B.S. in chemical engineering for graduate credit.

Mass and energy balances in batch continuous and open systems. Process thermodynamics. Cryogenics. Properties of substances and mixtures. Phase equilibria. Chemical reaction equilibrium. Chemical reactor kinetics. Process design orientation.

807. Transfer and Separation Processes

Summer. 5(7-0) B.S. with a major in chemistry, biochemistry, or a closely allied area. Mathematics through calculus. College level physics. General physical, and organic chemistry. Not open to students with B.S. in chemical engineering for graduate credit.

Momentum, energy, and mass transfer. Laminar and turbulent flow. Fluid friction. Dimensional analysis. Heat through stationary and flowing materials. Interchangers. Condensation. Boiling. Binary and multicomponent distillation, absorption, extraction.

811. Advanced Chemical Engineering Thermodynamics I

Fall. 3(3-0) CHE 311, CHE 411. CEM 361.

Advanced treatment of the laws of thermodynamics. Cryogenic processes. Corresponding state and higher parameters in computing properties of chemical compounds and solutions.

817. Advanced Chemical Reaction Engineering I

Spring. 3(3-0) CHE 428.

Treatment of absorption and catalysis and their application to catalytic reactors. Heat, momentum, and mass-transfer in fixed-bed and fluidized-bed reactors. Noncatalytic heterogenous reactions. Homogeneous chain reactions and free radical mechanisms. Computer applications to solution of complex kinetic problems.

826. Flow of Heat I

Spring. 3(3-0) CHE 343.

Steady and unsteady state heat transfer. Conduction and convection in flow and non-flow systems.

831. Advanced Distillation

Winter. 3(3-0) CHE 343.

Stagewise calculation in distillation processes. Computer techniques. Batch, continuous, binary and multi-component calculations. Tray hydrodynamics and efficiency. Process control and energy integration.

832. Advanced Absorption and Extraction

Sprtng. 3(3-0) CHE 343.

Mass transfer in absorption and extraction processes. Continuous and stagewise phase contactors. Column hydrodynamics and plate efficiency. Design and control principles.

850. Fluid Flow and Rheology

Fall. 3(3-0) CHE 481 or approval of department.

Application of fluid dynamics to chemical engineering systems. Balance principles for fluids; Newtonian and non-Newtonian behavior; theory and practice of laminar and turbulent flows; stability.

851. Mass Transfer

Winter. 3(3-0) CHE 850.

Formulation of component material balances; Fick's first and second laws; convective mass transfer; multicomponent fluxes; boundary layer theory and interfacial mass transfer for laminar and turbulent flows.

881. Seminar

Fall, Winter, Spring. 1(0-2) May reenroll for a maximum of 6 credits.

Detailed library investigation of one or more specialized aspects of chemical engineering, such as recent theoretical developments in one of the unit operations; presentations of these studies to a seminar group. Participation generally required each term of residence.

886. Selected Topics in Chemical Engineering

Fall, Winter, Spring, Summer. 3(3-0) May reenroll for a maximum of 9 credits if a different topic is taken.

A newly developing area of chemical engineering selected by the department for offering each term. Information on the specific topic to be covered should be obtained from the department office before registration.

888. Research Survey

Fall, Winter, Spring, Summer. 1 to 3 credits. May reenroll for a maximum of 3 credits.

Literature search, problem analysis, and layout of a complete research program.

893. Special Problems

Fall, Winter, Spring, Summer. 3(3-0) May reenroll for a maximum of 9 credits. Approval of department.

899. Master's Thesis Research

Fall, Winter, Spring, Summer. Variable credit. Approval of department.

912. Advanced Chemical Engineering Thermodynamics II

Spring of even-numbered years. 3(3-0) Approval of department.

Relation of thermodynamics to quantum theory and statistical mechanics. Computation of chemical engineering thermodynamic data from spectral measurements. Irreversible thermodynamics.

918. Advanced Chemical Reaction Engineering II

Fall of odd-numbered years. 3(3-0) Approval of department.

Quantitative treatment of current literature in chemical kinetics and reaction engineering.

999. Doctoral Dissertation Research

Fall, Winter, Spring, Summer. Variable credit. Approval of department.

151. Principles of Chemistry I
Fall, Winter. 4(4-0) MTH 108 or MTH 111 or concurrently; CEM 140 or satisfactory chemistry placement test score.
First of a 3-term sequence for science majors, chemical engineering students, and others desiring a comprehensive general chemistry sequence. Atomic and molecular structure; stoichiometry; solids, liquids, and gases; solutions.

152. Principles of Chemistry II
Winter, Spring. 3(3-0) MTH 112 or concurrently; CEM 151 or CEM 141A or CEM 141B or CEM 181H.
Continuation of CEM 151. Chemical thermodynamics; kinetics, acids, bases, and aqueous equilibria; electrochemistry.

153. Introductory Inorganic Chemistry
Fall, Spring. 3(3-0) CEM 152 or CEM 182H.
Continuation of CEM 152. Descriptive inorganic chemistry with further discussion of bonding.

161. Introductory Chemistry Laboratory
Fall, Winter, Spring, Summer. 1(0-3) CEM 140 or CEM 141A or CEM 141B or CEM 151 or concurrently.
Laboratory work in chemistry, including quantitative physiochemical or analytical experiments and chemical synthesis.

162. Quantitative Analysis
Fall, Winter, Spring, Summer. 3(1-6) CEM 141A or CEM 141B or CEM 151 or CEM 181H; CEM 161 or CEM 184H.
Laboratory work in quantitative chemistry.

163. Introductory Inorganic Laboratory
Spring. 2(0-6) CEM 142 or CEM 153 or concurrently; CEM 161.
Qualitative analysis and inorganic preparations.

181H. Honors Chemistry I—Principles
Fall. 4(4-0) An A average in high school chemistry, physics and mathematics; MTH 112 or MTH 122 concurrently. Results of examination during orientation; approval of department.
Subatomic, atomic and molecular structure; quantum theory and bonding; experimental methods of structure determination; states of matter; nuclear chemistry.

182H. Honors Chemistry II—Principles
Winter. 4(4-0) CEM 181H with grade of 3.0 or better and/or approval of department. MTH 113 or MTH 123 concurrently.
Kinetic theory of gases, thermodynamics, chemical equilibrium, electrochemistry, chemical kinetics, properties of solutions, macromolecular chemistry.

183H. Honors Chemistry III—Inorganic Chemistry
Spring. 3(3-0) CEM 182H with grade of 3.0 or better and/or approval of department.
Descriptive inorganic chemistry by periodic groups of elements. Nomenclature, bonding, sterochemistry, and reactions of compounds of the representative and transition elements.

184H. Honors Chemistry Laboratory I
Fall. 1(0-3) CEM 181H concurrently; approval of department.
Techniques of measurement; errors and significant figures; experiments related to atomic and molecular structure.

185H. Honors Chemistry Laboratory II
Winter. 2(0-6) CEM 184H; CEM 182H concurrently; approval of department.
Experiments related to gas behavior, thermodynamics, electro-chemistry, chemical kinetics and properties of solutions.

186H. Honors Chemistry Laboratory III
Spring. 2(0-6) Approval of department.
Introductory independent laboratory work in chemistry.

241. Organic Chemistry
Fall, Winter, Summer. 4(4-0) CEM 141A or CEM 141B or CEM 152 or CEM 181H; CEM 161 or CEM 184H.
Common classes of organic compounds with emphasis on nomenclature, structural principles, reactions and reaction mechanisms.

242. Organic Chemistry
Winter, Spring, Summer. 4(4-0) CEM 241.
Continuation of CEM 241 with emphasis on polyfunctional compounds, particularly groups of compounds having biological significance.

243. Organic Chemistry Laboratory
Fall, Winter: 1(0-2) Summer: 1(0-3). CEM 241 or concurrently.
Introduction to standard organic laboratory techniques.

244. Organic Chemistry Laboratory
Winter, Spring, Summer. 1(0-3) CEM 241, CEM 243, CEM 242 concurrently.
Organic preparations and qualitative analysis.

245. Organic Chemistry
Fall, Spring. 4(4-0) CEM 242.
Selected topics of organic chemistry, especially compounds of biological interest, discussed with emphasis on mechanisms and stereochemistry. Topics include polymers, amino acids, proteins, sugars, terpenes, steroids, and alkaloids.

333. Instrumental Methods
Spring. 4(2-6) CEM 143 or CEM 241 or CEM 351; CEM 162.
Principles, applications of separation and instrumental analysis. Atomic emission, absorption, fluoresence spectrometry; UV, visible, IR spectrophotometry; molecular fluoresence; gas and other chromotography; electro-analytical chemistry; electrophoresis; radiochemistry.

351. Organic Chemistry
Fall. 3(4-0) CEM 152 or CEM 182H.
A comprehensive introduction to the fundamentals of organic chemistry designed for chemistry majors but open to others who desire a rigorous, modern treatment of the subject.

352. Organic Chemistry
Winter. 3(4-0) CEM 351.
Continuation of CEM 351.

353. Organic Chemistry
Spring. 3(4-0) CEM 352.
Continuation of CEM 352.

354. Organic Chemistry Laboratory
Winter. 2(0-6) CEM 162 or CEM 185H; CEM 351.
A laboratory course in modern techniques of organic chemistry, including qualitative organic analysis.

355. Organic Chemistry Laboratory
Spring. 2(0-6) CEM 352, CEM 354.
Continuation of CEM 354.

356. Organic Chemistry Laboratory
Fall. 2(0-6) CEM 355.
Continuation of CEM 355.

361. Chemical Thermodynamics
Fall. 3(4-0) One year general chemistry; one year general physics; MTH 215.
Thermodynamics. Properties of gases. Laws of thermodynamics, properties of ideal and non-ideal solutions, thermodynamics of chemical reactions, activities in non-ionic systems.

362. Analytical-Physical Chemistry I
Winter. 3(4-0) CEM 361.
Applications of thermodynamics. Activity coefficients, ionic solutions, cell potentials, ionic equilibria including acid-base, complexation, solubility and redox equilibria, phase equilibria, distillation, extraction, chromatography.

363. Analytical-Physical Chemistry II
Spring. 3(4-0) CEM 362.
Chemical kinetics. Homogeneous kinetics, reaction mechanisms, temperature dependence of reaction rates, transport process, heterogeneous kinetics, electrode kinetics, X-ray diffraction, crystal structure.

372. Analytical-Physical Chemistry Laboratory I
Winter. 2(1-3) CEM 162; CEM 383, or CEM 361.
Measurement techniques. Temperature measurement and control, pressure, caloimetry, pH, acid-base titrations, cell potentials, treatment of data.

373. Analytical-Physical Chemistry Laboratory II
Spring. 2(1-3) CEM 372.
Instrumental measurements. Electrode potentials, chromatography, spectrophotometry, electrolytic conductance, solution kinetics.

383. Physical Chemistry: Introductory
Fall, Summer. 3(4-0) CEM 143 or CEM 241 or CEM 351; MTH 113.
Classical and chemical thermodynamics. Introduction to the laws and their applications in treating chemical reactions, pure substances, ideal and non-ideal mixtures, and colligative properties.

384. Physical Chemistry: Introductory
Spring. 3(4-0) CEM 383.
Atomic and molecular structure. Atomic and molecular orbitals and chemical bonding. Rotational, vibrational and electronic spectra, nuclear magnetic resonance and electron spin resonance.

385. Physical Chemistry: Introductory
Winter. 3(4-0) CEM 383.
Electrochemistry and electromotive force. Chemical kinetics. Macromolecules and biochemical systems. Nuclear chemistry.

400H. Honors Work
Fall, Winter, Spring, Summer. 1 to 3 credits. May reenroll for a maximum of 18 credits. Seniors, approval of department.
Assigned reading and investigation in chemistry under the supervision of the staff. The program will include some creative work.

411. Inorganic Chemistry I

Fall. 3(3-0) CEM 385 or CEM 363.

Principles of structure and bonding in inorganic chemistry, crystal symmetry, coordination chemistry, solvent systems, hydrogen bonding and selected examples from the chemistry of various elements.

412. Inorganic Chemistry II

Winter. 3(3-0) CEM 411.

Inorganic chemistry viewed in a variety of examples of reactions, structure, mechanisms, etc., from representative main group elements and transition elements.

419. Problems and Reports

Fall, Winter, Spring, Summer. 1 to 8 credits. May reenroll for a maximum of 8 credits. Approval of department.

430. Introduction to Radioactivity and Radioisotope Techniques

Spring, Summer. 2(3-0) or 3(3-0) One year each of general college chemistry and physics. Interdepartmental with and administered by Physics.

First 7 weeks. Elementary nuclear processes and properties with emphasis on radioactivity, its measurement, and its interaction with matter. Effects of radiation on chemical and biological systems. Applications of nuclear technology, safety and environmental factors.

Last 3 weeks. Fundamentals of nuclear models, reactions and decay mechanisms. Basic principles of nuclear reactors and accelerators.

431. Laboratory for Radioactivity and Radioisotope Techniques

Spring, Summer. 1(0-3) CEM 161, PHY 430, concurrently. CEM 162 recommended. Interdepartmental with and administered by Physics.

Introduction to nuclear instrumentation. Experimental techniques for application of radioisotopes to problems in chemistry, the life sciences, and industry.

461. Theoretical Chemistry I

Fall. 3(4-0) One year general chemistry; one year general physics; MTH 215.

Quantum chemistry. Wave properties, postulates of quantum mechanics, hyrdodgen atom, helium atom, orbital theories, ionic bonds, simple molecules, valence-bond and molecular-orbital theories, complex molecules, introduction to spectra.

462. Theoretical Chemistry II

Winter. 3(4-0) CEM 361, CEM 461.

Spectroscopy and molecular structure. Electronic, infrared, Raman, and microwave spectroscopy, magnetic susceptibility and magnetic resonance, statistical mechanics, statistical thermodynamics, kinetic theory of gases, absolute rate theory.

471. Analytical-Physical Chemistry Laboratory III

Fall. 2(0-6) CEM 363, CEM 373.

Kinetics, operational amplifiers, polarography, coulometry, electrochemical kinetics, stopped-flow kinetics, digital measurements, neutron activation.

472. Analytical-Physical Chemistry Laboratory IV

Winter. 2(0-6) CEM 461, CEM 471.

Molecular properties. Mass spectrometry, nuclear and electron spin resonance spectroscopy, infrared spectroscopy, dipole moments, magnetic susceptibility, gaseous decomposition kinetics.

484. Modern Physical Chemistry

Spring. 3(3-0) May reenroll for a maximum of 6 credits if a different topic is taken. CEM 462.

Topics may be selected from the following: physical properties and structure, molecular structure, spectroscopy, theory of solutions.

499. Seminar on Chemical Physics

Fall, Winter, Spring. 1(1-0) May reenroll for a maximum of 3 credits. One year of analytical-physical chemistry. MTH 215; PHY 428.

Literature of chemical physics through oral reports on selected journal articles in the area.

810. Advanced Inorganic Chemistry

Fall. 3(3-0) Approval of department.

Structure, bonding, and reactivity patterns of inorganic compounds, with emphasis on nonmetallic elements and reactions in nonaqueous media.

811. Symmetry, Group and MO Theory

Winter. 3(3-0) Approval of department.

Applications of group and molecular orbital theory to chemical bonding, structure and reactions.

813. Advanced Inorganic Chemistry—Metals

Spring. 3(3-0) CEM 811.

Continuation of CEM 811 with emphasis on the structure and chemistry of the metals.

833. Analytical Spectroscopy

Spring. 3(3-0) Approval of department.

834. Advanced Analytical Chemistry

Fall. 3(3-0) Approval of department.

Consideration of principles and equilibria pertaining to aqueous and non-aqueous neutralization, redox and complexation reactions and the various separation techniques employed in analyses.

835. Spectrochemical Methods of Analysis

Winter. 3(3-0) or 4(3-4) Approval of department.

Principles and applications of atomic absorption, emission, fluorescence; arc and spark emission spectroscopy; UV, visible, IR spectrophotometry; spectrophotometric titrations; reaction rate methods; molecular fluorescence, phosphorescence spectrometry; other optical spectrometric methods.

836. Separations

Spring of odd-numbered years. 3(3-0) Approval of department.

Physical and chemical methods of separation.

837. Electroanalytical Chemistry

Spring of even-numbered years. 3(3-0) Approval of department.

Theory and applications of modern electroanalytical chemistry to chemical and biomedical problems. Coulometry, electrometric titrations, ion-selective voltammetry; electrochemical synthesis and preparation of species for spectroscopy; trace analysis.

838. Scientific Instrumentation

Fall, Spring, Summer. 3(1-6) May reenroll for a maximum of 9 credits. Approval of department.

Scientific measurements. Principles and applications of servo systems, operational amplifiers, linear and digital solid state devices, analog, digital and hybrid instrumentation systems, and minicomputers for scientific measurements.

844. Structural Elucidation by Instrumental Methods

Fall. 3(3-0) Approval of department.

A practical instrumental analysis course with the major emphasis on the interpretation of data rather than a detailed description of the instrumentation. The fundamental principles behind the various measurements will be discussed in a general way, and important instrumental limitations will be noted.

860. Organic Reactions: A Mechanistic Approach

Fall. 3(3-0) CEM 353; CEM 462 or approval of department.

Organic reactions are presented in a mechanistic framework. Reactions which proceed via carboncations, carbanions, free radicals, carbenes, arynes and other reactive intermediates, and concerted reactions are included.

861. Structure of Organic Compounds

Winter. 3(3-0) CEM 860 or approval of department.

Structural and stereochemical principles will be developed and illustrated. Spectroscopic data will be used to illustrate the principles and to determine structure, with an emphasis on nuclear magnetic resonance spectroscopy.

862. Advanced Synthetic Organic Chemistry

Spring. 3(3-0) CEM 860 or approval of department.

The strategy and methods of organic synthesis will be discussed.

880. Atomic and Molecular Structure

Fall. 3(3-0) CEM 462 or approval of department.

Basic concepts of non-relativistic quantum mechanics will be developed and employed in a description of atomic and molecular structure.

881. Thermodynamics

Winter. 3(3-0) Approval of department.

Laws of thermodynamics and their application to pure substances and solutions.

883. Chemical Kinetics

Spring. 3(3-0) CEM 880.

Rates and mechanisms of chemical reactions, reaction rate theory, kinetic theory of gases, photochemistry.

890. Graduate Problems and Reports

Fall, Winter, Spring, Summer. 1 to 6 credits. May reenroll for a maximum of 12 credits. Approval of department.

899. Master's Thesis Research

Fall, Winter, Spring, Summer. Variable credit. Approval of department.

Research in inorganic, analytical, organic, and physical chemistry.

913. Selected Topics in Inorganic Chemistry

Fall, Spring, 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken.

Rare earth elements; recent advances in the chemistry of metals or nonmetals, high-temperature chemistry. Coordination chemistry and nonaqueous solvents.

918. Seminar in Inorganic Chemistry

Fall, Winter, Spring. 1(2-0) May reenroll for a maximum of 3 credits.

Discussions of recent advances and reports by graduate students on research problems.

924. Selected Topics in Analytical Chemistry

Fall, Winter, Spring. 3(3-0) or 2(2-0) May reenroll for a maximum of 9 credits if different topic is taken.

Among topics which may be discussed are: advances in electroanalytical chemistry or spectroscopy; nonaqueous solvents; complexation equilibria; surface chemistry; analytical chemistry of polymers.

938. Seminar in Analytical Chemistry

Fall, Winter, Spring. 1(1-0) May reenroll for a maximum of 3 credits.

Discussions of recent advances and reports by graduate students on research problems.

956. Selected Topics in Organic Chemistry

Fall, Winter, Spring. 2(2-0) or 3(3-0) May reenroll for a maximum of 12 credits if different topic is taken. Approval of department.

Topics may be selected from heterocyclic chemistry, natural products, free radicals, carbonium ions, organic sulfur or nitrogen compounds, acidity functions, isotope effects, photochemistry and others.

958. Seminar in Organic Chemistry

Fall, Winter, Spring. 1(2-0) May reenroll for a maximum of 3 credits.

Discussions of recent advances and reports by graduate students on research problems.

985. Statistical Thermodynamics

Winter, Spring. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Partition functions, spectroscopic measurements and thermodynamic applications. Nonequilibrium statistical mechanics and thermodynamics. Time correlation functions and spectroscopic lineshapes, light scattering, and magnetic relaxation. Transport properties of fluids and gases.

987. Selected Topics in Physical Chemistry

Fall, Winter. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Mathematical preparation for quantum chemistry. Selected topics as: kinetics and photochemistry, macromolecular and surface chemistry, molecular spectroscopy, electro and magnetic properties of matter, application of statistical mechanics to chemical problems.

988. Selected Topics in Physical Chemistry

Winter, Spring. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Topics may be chosen from analysis and interpretation of the spectra of molecules, advanced molecular structure, magnetic resonance, spectroscopy, X-rays and crystal structure, statistical mechanics.

991. Selected Topics in Quantum Chemistry

Fall, Winter. 3(3-0) May reenroll for a maximum of 9 credits if different topic is taken. Approval of department.

Principles of quantum mechanics and application to chemical problems. Selected topics from spectroscopy, properties of atoms and molecules in electric and magnetic fields, and theories of molecular electronic structure.

998. Seminar in Physical Chemistry

Fall, Winter, Spring. 1(1-0) May reenroll for a maximum of 3 credits.

Discussions of recent advances and reports by graduate students on research problems.

999. Doctoral Dissertation Research

Fall, Winter, Spring, Summer. Variable credit. Approval of department.

Research in analytical, inorganic, organic, and physical chemistry.

CHINESE

See Linguistics and Germanic, Slavic, Asian and African Languages.

CIVIL AND ENVIRONMENTAL ENGINEERING

(Name change effective July 1, 1985. Formerly the Department of Civil and Sanitary Engineering.)

College of Engineering

Civil Engineering — C E

251. Elementary Surveying

Spring. 4(3-3) Not open to majors.

Use of the tape, compass, level, and transit with simple maps; traverse closure and area computations. Profile, cross section and stadia surveys, U.S. land system.

252. Surveying I

Fall, Spring. 5(4-3) Engineering majors or approval of department.

Instruments, theory of measurements, error analysis, stadia, horizontal and vertical curves, U.S. Public Land System, observation for meridian.

280. Introduction to Environmental Engineering

Fall, Winter, Spring. 4(4-0) CEM 141, or CEM 131, MTH 112, CPS 120.

Hydrology; ground water and surface water supply systems; wastewater treatment, methods of pollution control for solid waste, air, and noise.

305. Structural Analysis I

Fall, Winter. 3(3-0) MMM 211.

Stability and determinacy; linearity. Plane trusses; shear and bending in beams and frames. Virtual work calculation of forces and displacements in statically-determinate plane structures.

306. Structural Analysis II

Winter, Spring. 3(3-0) C E 305.

Beam-columns. Flexibility method for plane structures. Kinematic indeterminacy. Stiffness method for plane structures. Moment distribution method. Design loading of structures.

308. Engineering Materials I

Fall, Winter, Spring. 4(3-3) MMM 211 or concurrently.

Structure; composition; physical, mechanical and rheological properties of non-metallic construction materials. Emphasis on aggregates, asphalt, inorganic cements, concrete, and wood.

312. Soil Mechanics

Winter, Spring, Summer. 4(3-3) MMM 211.

Engineering properties of soils and their measurement. Effective stress concept; permeability; fluid flow in soils; stress-strain behavior; soil strength; compaction and consolidation of soils; field exploration and design problems.

321. Introductory Fluid Mechanics

Fall, Winter. 4(3-2) MTH 310.

Fluid properties; hydrostatics; control volume aproach to conservation of mass, momentum and energy; dimensional analysis and dynamic similitude; fluid resistance; pipe and open channel flows; boundary layer concepts.

346. Transportation

Winter, Spring, Summer. 4(4-0) MTH 113.

Planning, design, and evaluation of transportation systems; highway, street, and intersection capacity; basic elements of geometric design.

347. Geometric Design of Highways

Fall, Winter. 4(3-3) C E 346 or concurrently.

Geometric design of streets and highways as related to capacity, construction costs, and safety. State and national design standards and practice.

370. Cost and Optimization Engineering

Fall, Winter. 3(3-0) MTH 113.

Formulation of engineering decisions governed by current and future costs and returns. Comparison and optimization of alternative engineering projects, products and processes.

372. Construction Estimating

Fall, Spring., 3(3-0) Juniors.

Cost studies of contruction activities with emphasis on labor productivity and operating characteristics of equipment under various site conditions. Interpretation of drawings and specifications.

374. Legal Aspects of Engineering

Spring. 3(3-0) Juniors.

The professional engineer's relationship with the legal aspects of engineering. Special emphasis on contract documents.

390. Civil Engineering Analysis

Fall, Spring. 3(2-3) MTH 310, CPS 120.

Analysis of civil engineering problems by numerical methods. Use of microcomputers to analyze problems. Technical reports to present methods and computed results.